

Annual Drinking Water Quality Report

Monitoring Performed January - December 2024

Central Elmore Water & Sewer Authority

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Central Elmore Water & Sewer Authority maintains and operates a 12 million gallon per day surface water treatment plant at our primary water source on Lake Martin.



Here at CEW&SA, we serve approximately 13,238 customers of our own; along with Rockford Utilities (1,371 customers), Eclectic Water Works & Sewer Department (1,736 customers), Friendship Water Works (1,370 customers), and Wetumpka Water Works & Sewer Board (3,400 customers).

Each customer refers to a meter served, which translates into approximately 63,345 persons CEW&SA serves.

Our territory covers approximately 350 square miles out of the 657 square miles contained in Elmore County. We currently maintain over 790 miles of water mains in our territory along with 12 water storage facilities holding a total of almost 7.7 million gallons.

We want our valued customers to be informed about their water utility.

Regularly scheduled Board Meetings are held the third Tuesday of each
month at the main office located at: 716 US Highway 231.

Board of Directors

Fred Braswell, III - Chairman
Bill Newton - Vice Chairman
Conrad White - Director
Chad Shaw - General Manager
Tina Stanley - Office Manager

Monitoring Schedule

Our water sources are routinely monitored for contaminants, according to a schedule determined by Federal and State regulations. Every water system has individually assigned monitoring requirements.

ADEM allows monitoring for some contaminants less than once per year because the concentrations of these contaminants do not change frequently. The following table shows the most recent year and the next monitoring requirement for the contaminant groups.

| Contaminant Monitored | Date Monitored / Next Monitoring |
|--|-------------------------------------|
| Inorganic Contaminants | Annually |
| Lead/Copper | 2022 / 2025 |
| Microbiological Contaminants | Monthly |
| Nitrates | Annually |
| PFAS | Quarterly |
| Radioactive Contaminants | 2022 / 2031 |
| Synthetic Organic Contaminants (including pesticides and herbicides) | 2022 / 2025 |
| Volatile Organic Contaminants | Annually |
| Disinfection By-products | Quarterly |

Variances and Exemptions

ADEM or the EPA can give permission not to meet an MCL or a treatment technique under certain conditions.

Based on a study conducted by ADEM with the approval of the EPA, a statewide waiver for the monitoring of asbestos and dioxin was issued.

Thus, monitoring for these contaminants was not required.

It is an honor to once again present to you this Annual Water Quality Report. This year's report is an overview of 2024's water quality. The report has been prepared to meet the requirements of the 1996 Safe Drinking Water Act (SDWA) adopted by Congress and to provide our customers with information about their water system. It has been the goal of Management to become more transparent for the customers. Informed customers are our biggest allies.

As part of the new EPA regulations on lead and copper, CEW&SA staff built a database that includes all water service material on both sides of the meter within our service territory. I am proud to say that no lead service lines were discovered during this process. We were able to submit the completed database to ADEM and EPA before the October 2024 deadline.

The EPA released new PFAS regulations in April 2024. We are proud to say that our quarterly sample results continue to be below the new EPA limits. The water provided to you by Central Elmore Water & Sewer Authority (CEW&SA) continues to meet or exceed all state and federal water quality regulations. CEW&SA has never had a violation of contamination levels in the water we supply you, our valued customers. Go to our website and Facebook page for our latest news release on the EPA's PFAS regulations.

In 2024, Management introduced a Capital Improvement Plan (CIP) to combat the leaks and water loss attributed to aging infrastructure. Funding for the 2025 projects was tied directly to proper investing. Moving forward, the Board and management have expressed the desire to fund projects in the CIP each year and will do so as funding is available, which is a testament to properly managing costs and appropriating the savings accordingly.

Also in 2024, CEW&SA's consulting engineers began developing construction plans for a Granular Activated Carbon System. After completing a two-year pilot study, this system was designed to remove the Geosmin and MIB associated with the taste and odor issues caused by the algae in Lake Martin. The overall project is designed to include the replacement of outdated plant PLCs, the Hypo Generation System, and the filter bed media. The new system is expected to become operational in 2026.

For the first time in several years, CEW&SA's employment numbers are back to normal. The Filter Plant hired two Grade 4 Operators and a Maintenance Technician along with several Field Service Technicians in Operations. CEW&SA replaced its long-time Accountant after her many years of faithful employment and retirement. CEW&SA continues to operate with the same number of employees since before 2010.

I encourage you to take the time to review this report. If you have any questions concerning this report or CEW&SA, please contact me, Chad Shaw, General Manager, at 334-567-6814, Monday - Friday, 7:30 a.m. to 4:30 p.m. and I will be glad to address any concerns you may have.

Chadwick E. Shaw, P.E. General Manager

General Information Regarding Drinking Water Contaminants

All drinking water, including bottled drinking water, may be reasonably expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the Environmental Protection Agency's (EPA) Safe Drinking Water Hotline at 800-426-4791.

In order to ensure that tap water is safe to drink, EPA prescribes regulations that limit the amount of certain contaminants in water provided by public water systems. Food and Drug Administration (FDA) regulations establish limits for contaminants in bottled water.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and radioactive material, and can pick up substances resulting from the presence of animals or from human activity. Contaminants that may be present in source water include:

- Microbial contaminants, such as viruses and bacteria, may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- Inorganic contaminants, such as salts and metals, can be naturally occurring or result from urban stormwater run-off, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- Pesticides and herbicides may come from a variety of sources such as agriculture, stormwater run-off, and residential uses.
- Organic chemical contaminants, including synthetic and volatile organic chemicals, are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, and septic systems.
- Radioactive contaminants, can be naturally occurring or be the result of oil and gas production and mining activities.

Some people may be more vulnerable to contaminants in drinking water than the general population. People who are immuno-compromised such as cancer patients undergoing chemotherapy, organ transplants recipients, people with HIV/AIDS positive or other immune system disorders, some elderly, and infants can be particularly at risk from infections. People at risk should seek advice about drinking water from their healthcare providers. For people who may be immuno-compromised, a guidance document developed jointly by the Environmental Protection Agency and the Center for Disease Control (CDC) is available online online www.epa.gov/safewater or by calling the Safe Drinking Water Hotline (800-426-4791).

Water systems also test your source water for pathogens, such as Cryptosporidium and Giardia. These pathogens can enter the water from animal or human waste. EPA/CDC guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbiological contaminants can also be obtained by calling the hotline or online www.epa.gov/safewater.

Important Health Information about Lead

Lead can cause serious health effects in people of all ages, especially pregnant people, infants (both formula-fed and breastfed), and young children. Exposure to lead in drinking water can cause serious health effects in all age groups, especially for pregnant women and young children. Infants and children can have decreases in IQ and attention span. Lead exposure can lead to new learning and behavior problems or exacerbate existing learning and behavior problems. The children of women who are exposed to lead before or during pregnancy can have increased risk of these adverse health effects. Adults can have increased risks of heart disease, high blood pressure, kidney, or nervous system problems.

Lead in drinking water is primarily from materials and parts used in service lines and home plumbing. Central Elmore Water & Sewer Authority is responsible for providing high-quality drinking water and removing lead pipes but cannot control the variety of materials used in the plumbing in your home. Because lead levels may vary over time, lead exposure is possible even when your tap sampling results do not detect lead at one point in time.

Lead levels in your drinking water are likely to be higher if:

- Your home or water system has lead pipes, or
- Your home has faucets or fittings made of brass which contains some lead, or
- · Your home has copper pipes with lead solder and you have naturally soft water, and
- · Water often sits in the pipes for several hours

You can help protect yourself and your family by identifying and removing lead materials within your home plumbing and taking steps to reduce your family's risk:

- Using a filter, certified by an American National Standards Institute accredited certifier to reduce lead, is effective in reducing lead exposures. Follow the instructions provided with the filter to ensure the filter is used properly.
- Clean your aerator. Regularly clean your faucet's screen (also known as an aerator). Sediment, debris, and lead particles can collect in your aerator. If lead particles are caught in the aerator, lead can get into your water.
- Use only cold water for drinking, cooking, and making baby formula.
 - o Boiling water does not remove lead from water.
- Before using tap water for drinking, cooking, or making baby formula, flush your pipes for several minutes.
 - You can do this by running your tap, taking a shower, doing laundry or a load of dishes.
 - If you have a lead service line or galvanized requiring replacement service line, you may need to flush your pipes for a longer period.

If you are concerned about lead in your water, you may wish to have your water tested, contact Central Elmore Water & Sewer Authority at (334) 567-6814.

Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water hotline or at www.epa.gov/safewater/lead.



During the past year, we have taken thousands of water samples in order to determine the presence of any primary, secondary, or unregulated contaminants. The water quality information presented in the tables below is from the most recent monitoring periods for each group. These tables only include those contaminants that were detected in the water.

| | | Table of D | etected Prim | ary Contaminan | ts | |
|---|--|-------------------------------|-------------------|--------------------------------------|--------------|---|
| Primary Standards - Manda | tory standards se | t by the Safe E | rinking Water Act | used to protect public | health. Thes | e apply to all public water systems. |
| Contaminant & Unit of MSMT | MCL, TT, or MRDL (What's Allowed?) | MCLG (What's the Goal?) | Max Detected | Range of Detected Low - High (MD) | Violation | Major Sources |
| | | BACTER | IOLOGICAL CONT | 'AMINANTS - 2024 | | |
| Total Organic Carbon TOC (ppm) | П | NA NA | 1.48 | ND - 1.48 þ | No | Naturally present in the environment |
| | | | | | | |
| Arsenic (ppb) | 0.010 | 0 | 0.39 | 0.39 | No | Erosion of natural deposits; runoff from orchards, runoff from glass and electronics production wastes |
| Barium (ppm) | ž | 2 | 0.0115 | 0.0115 | No | Discharge of drilling wastes; Discharge from mo refineries; Erosion of natural deposits |
| Chromium (ppb) | 100 | 100 | 0.598 | 0.598 | No | Discharge from steel and pulp mills; Erosion on natural deposits |
| Fluoride (ppm) | 4 | 4 | 0.901 | 0.901 | No | Water additive which promotes strong teeth; ero of natural deposits; Discharge from fertilizer a aluminum factories |
| Mercury (ppb) | 2 | ND | 0.44 | 0.44 | No | Corrosion of household plumbing systems; Eros of natural deposits |
| | | LEAD | & COPPER (TAP | WATER) - 2022 | | |
| Copper-action level at consumer taps (ppm) | AL=1.3 | 1.3 | 0.0867 | 0.0082 - 0.0867 | No | Corrosion of household plumbing systems; Eros of natural deposits |
| Lead - action level at consumer taps (ppb) | AL=15 | 0 | 1.4 | ND-1.4 | No | Corrosion of household plumbing systems; Eros of natural deposits |
| | | | | | | |
| Total Haloacetic Acids HAA (ppb) | 60 | NA | 24.9 | LRAA Range 8.8 - 21.5 | No | By-product of drinking water disinfection |
| Total Trihalomethanes TTHM (ppb) | 80 | NA | 50.4 | LRAA Range 14.8 - 35.0 | No | By-product of drinking water disinfection |

b The percentage of Total Operatic Carbon (TOC) removal was measured each month and the system met all TOC removal requirements set

There is convincing evidence that the addition of a disinfectant is necessary for the control of microbial contains.

| | | | | table are from 2024 |
|---|--------------------------------------|---------|---------------------|---|
| | Contaminant & Unit of MSMT | MCL | Maximum Detected | Major Sources |
| | Chloride (ppm) | 250 | 11.6 | Naturally occurring in the environment or as a result of agricultural runoff |
| | Copper (ppm) | 1.0 | 20.1 | Erosion of natural deposits; Corrosion of household plumbing systems |
| Г | Manganese (ppm) | 0.05 | 2.4 | Erosion of natural deposits; Leaching from pipes |
| | pH (std units) | 6.5-8.5 | 7.7 | Naturally occurring in the environment or as a result of treatment with water additives |
| Г | Sulfate (ppm) | 250 | 12.8 | Naturally occurring in the environment or as a result of industrial discharge or as a result of agricultural runoff |
| | Total Dissolved Solids (ppm) | 500 | 34 | Naturally occurring in the environment or as a result of industrial discharge or as a result of agricultural runoff |
| | Zinc (ppm) | 5 | 2.5 | Erosion of natural deposits; Discharge from refineries and factories; Runoff from landfills |
| | Alkalinity, Total (as CA, Co3) (ppm) | NA. | 20 | Naturally occurring in the environment |
| Г | Calcium, as Ca (ppm) | NA. | 2.7 | Erosion of natural deposits |
| Г | Carbon Dioxide (ppm) | NA. | 17.5 | Erosion of natural deposits |
| Г | Conductivity (umhos) | NA. | 105 | Naturally occurring in the environment or as a result of treatment with water additives |
| Г | Hardness (ppm) | NA. | 11.3 | Naturally occurring in the environment or as a result of treatment with water additives |
| | Magnesium (ppm) | NA. | 1.12 | Erosion of natural deposits |
| Г | Nickel (ppm) | NA. | 0.0011 | Result of discharge by power plants, metal factories and waste incinerators or as a result of agricultural runoff |
| | Sodium (ppm) | NA. | 16.6 | Naturally occurring in the environment |

| Filter Plant 2024 Daily Testing | Range Low - High (MD) |
|------------------------------------|--------------------------|
| BACTERIOLOGICAL | CONTAMINANTS |
| Turbidity (NTU) E | 0.01 - 0.08 |
| INORGANIC CO | NTAMINANTS |
| Fluoride (ppm) | 0.30 - 0.80 |
| DISINFECTANTS & DISINI | FECTION BYPRODUCTS |
| Chlorine (ppm) | 1.6 - 2.1 |
| Chlorine Dioxide (ppb) | 0.06 - 0.32 |
| Chlorite (ppm) | 0.29 - 0.76 |
| SECONDARY & ADDITIO | NAL CONTAMINANTS |
| Alkalinity | 13-30 |
| Hardness | 9-22 |
| Iron | ND-0.09 |
| Manganese | ND-0.08 |
| pH | 7.3 - 8.0 |

£ Turbitidy is a measure of the cloudiness of the water We monitor it because it is a good indicator of water quality. High turbidity can hinder the effectiveness of disinfectance

0.82 - 1.2

| | Unregulated (| Contaminants - | - 2024 |
|-------------------------------|---------------------|----------------------|--------------------------------------|
| Contaminant & Unit of MSMT | Average Detected | Range of Detected | Major Sources |
| Bromodichloromethane (ppb) | 6.69 | ND - 21.7 | Naturally occurring in the |
| Bromoform (ppb) | 0.82 | ND - 4.6 | environment or as a result of |
| Chioroform (ppb) | 17.0 | ND - 42.8 | industrial discharge or agricultural |
| Dibromochloromethane (ppb) | 0.16 | ND - 1.1 | runoff; by product of chlorination |



1 drop in a tanker truck = 1 pp OR, in terms of time, ppm can be thought of 32 years

| b | + |
|------------------|-----|
| as one second in | Pe |
| | cor |

Nitrite (measured as Nitrogen) NO2 (nom)

Not Applicable (NA)

in excess of 5 NTU is just noticeable to the average person.

ppb (parts per billion): micrograms per liter (µg/L) ppm (parts per million): milligrams per liter (mg/L)

ppt (parts per trillion): nanogram per liter (ng/L) pCi/L (picocuries per liter): a measure of radioactivity in water.

that still yields a just detectable odo

contaminant in drinking water.

Nephelometric Turbidity Unit (NTU): A measure of the clarity of the water. Turbidity

Threshold Odor Number (TON): The greatest dilution of a sample with odor-free water

Treatment Technique (TT): A required process intended to reduce the level of a

In order to ensure that tap water is safe to drink, the Environmental Protection Agency (EPA) prescribes regulations that limit the amount of contaminants in water provided by public water systems. All sources of drinking water contain some naturally occurring contaminants. At low levels, these substances are generally not harmful to our drinking water. Removing all contaminants would be extremely expensive, and in most cases, would not provide increased protection for public health. A few naturally occurring minerals may actually improve the taste of drinking water and have nutritional value at low levels.

| | | | | Table of F | Primary Co | ontaminants | | | | | |
|--|---------------------------------------|--------------|------------------------------------|---------------------------------------|-----------------|----------------------------------|---------------------------------------|-----------------|----------------------------------|---------------------------------------|-----------------|
| Contaminant & Unit of MSMT | MCL, TT, or MRDL (What's Allowed?) | Max Detected | Contaminant | MCL, TT, or MRDL (What's Allowed?) | Max Detected | Contaminant | MCL, TT, or MRDL (What's Allowed?) | Max Detected | Contaminant | MCL, TT, or MRDL (What's Allowed?) | Max Detected |
| BACTERIOLOGICAL | LCONTAMINANTS | | | | | | | | | | |
| Total Coliform Bacteria | < 5% present/absent | Absent | 1,1,1-Trichloroethane (ppb) | 200 | ND | Dalapon (ppb) | 200 | ND | Lindane (ppt) | 200 | ND |
| Fecal Coliform & E. coli | present/absent | Absent | 1,1,2-Trichloroethane (ppb) | 5 | ND | Dibromochloropropane (ppt) | 200 | ND | Methoxychlor (ppb) | 40 | ND |
| Total Organic Carbon (TOC) | TT | 1.48 | 1,1-Dichloroethylene (ppb) | 7 | ND | Di (2-ethylhexyl)adipate (ppb) | 400 | ND | o-Dichlorobenzene (ppb) | 600 | ND |
| Turbidity (NTU) | TT | 0.08 | 1,2,4-Trichlorobenzene (ppb) | 0.07 | ND | Di (2-ethylhexyl)phthalate (ppb) | 6 | ND | Oxamyl [Vydate] (ppb) | 200 | ND |
| RADIOLOGICAL (| CONTAMINANTS | | 1,2-Dichloroethane (ppb) | 5 | ND | Dinoseb (ppb) | 7 | ND | p-Dichlorobenzene (ppb) | 75 | ND |
| Beta/photon emitters (mrem/yr) | 4 | ND | 1,2-Dichloropropane (ppb) | 5 | ND | Dioxin [2,3,7,8-TCDD] (ppq) | 30 | NA | Pentachlorophenol (ppb) | 1 | ND |
| Alpha emitters (pCi/L) | 15 | ND | 2,4,5-TP [Silvex] (ppb) | 50 | ND | Diquat (ppb) | 20 | ND | Picloram (ppb) | 500 | ND |
| Combined radium (pCi/L) | 5 | ND | 2,4·D (ppb) | 70 | ND | Endothall (ppb) | 100 | ND | Polychlorinated biphenyls (ppt) | 0.5 | ND |
| INORGANIC CO | NTAMINANTS | | Acrylamide (ppb) | П | ND | Endrin (ppb) | 2 | ND | Simazine (ppb) | 4 | ND |
| Antimony (ppb) | 6 | ND | Alachlor (ppb) | 2 | ND | Epichlorohydrin (ppb) | п | ND | Styrene (ppb) | 100 | ND |
| Arsenic (ppb) | 10 | 0.39 | Atrazine (ppb) | 3 | ND | Ethylbenzene (ppb) | 700 | ND | Tetrachloroethylene (ppb) | 5 | ND |
| Asbestos (MFL) | 7 | NA NA | Benzene (ppb) | 5 | ND | Ethylene Dibromide (ppt) | 50 | ND | Toluene (ppm) | 1 | ND |
| Barium (ppm) | 2 | 0.0115 | Benzo(a)pyrene [PAHs] nanograms/L) | 200 | ND | Glyphosate (ppb) | 700 | ND | Toxaphene (ppb) | 3 | ND |
| Beryllium (ppb) | 4 | ND | Carbofuran (ppb) | 40 | ND | Heptachlor (ppt) | 400 | ND | trans-1,2-Dichloroethylene (ppb) | 100 | ND |
| Cadmium (ppb) | 5 | ND | Carbon Tetrachloride (ppb) | 5 | ND | Heptachlor Epoxide (ppt) | 200 | ND | Trichloroethylene (ppb) | 5 | ND |
| Chromium (ppb) | 100 | 0.598 | Chlordane (ppb) | 2 | ND | Hexachlorobenzene (ppb) | 1 | ND | Vinyl Chloride (ppb) | 2 | ND |
| Copper - action level at | | | Chlorobenzene (ppb) | 100 | ND | Hexachlorocyclopentadiene (ppb) | 50 | ND | Xylenes (ppm) | 10 | ND |
| consumer taps (ppm) | AL=1.3 | 0.0867 | cis-1,2-Dichloroethylene (ppb) | 70 | ND | | | | | | |
| Cyanide (ppb) | 200 | ND | | | | | | | | | |
| Fluoride (ppm) | 4 | 0.901 | Bromate (ppb) | 10 | ND | Chlorine Dioxide (ppb) | 800 | 0.32 | Total Haloacetic Acids HAA (ppb) | 60 | 24.9 |
| Lead - action level at | AL=15 | 1.4 | Chloramines (ppm) | 4 | ND | Chlorite (ppm) | 1 | 0.76 | Total Trihalomethanes TTHM (ppb) | 80 | 50.4 |
| consumer taps (ppb) | | | Chlorine (ppm) | 4 | 2.1 | | | | | | |
| Mercury (ppb) | 2 | 0.44 | | | | | | | | | |
| Nitrate (measured as Nitrogen) NO3 (ppm) | 10 | ND ND | CECONDARY C- ADD | ITIONIAL CONTA | STREET, STREET, | | | LIBIT | ECHI ATED CONTAMINANTO | | |

| | Thallium (ppb) | 2 | ND | & Unit of MSMT | (What's Allowed?) | Detected |
|---|--|------------------------|--------------------------|--------------------------------------|-------------------|----------|
| | | - | | Aluminum (ppm) | 0.05 to 0.2 | ND |
| | Abbreviation | ns & Definition | s | Chloride (ppm) | 250 | 11.6 |
| | | | | Color (color units) | 15 | ND |
| | Action Level (AL): The concentration of a | | ggers treatment or other | Copper (ppm) | 1.0 | 20.1 |
| | equirements that a water system shall fo | | | Corresivity | Non-corrosive | ND |
| | owest Running Annual Average (LRAA) | | | Fluoride (ppm) | 2.0 | ND |
| | amples taken at a particular monitoring warters. | location during the p | revious tour calendar | Foaming agents MBAS (ppm) | 0.5 | 2.4 |
| | Maximum Contaminant Level (MCL): The | highest contaminar | it level allowed in | Iron (ppm) | 0.3 | ND |
| | lrinking water. MCLs are set as close to th | | | Manganese (ppm) | 0.05 | 7.7 |
| t | reatment technology. | | | Odor (threshold odor number) | 3 | ND |
| | Aaximum Contaminant Level Goal (MCL | G): The level of a con | taminant in drinking | pH (std units) | 6.5 - 8.5 | 12.8 |
| | vater below which there is no known or e | xpected risk to healtl | 1. MCLGs allow for a | Silver (pprn) | 0.1 | 34 |
| | nargin of safety. | | | Sulfate (ppm) | 250 | 2.5 |
| | Maximum Detected (MD) | ammer W. I. I. I. | | Total Dissolved Solids (ppm) | 500 | 155 |
| | Maximum Residual Disinfectant Level (N Howed in drinking water, There is convin | | | Zinc (ppm) | 5 | 0.0019 |
| | lisinfectant is necessary for the control of | | | Alkalinity, Total (as CA, Co3) (ppm) | NA NA | 20 |
| | Aaximum Residual Disinfection Level G | | | Calcium, as Ca (ppm) | NA NA | 2.7 |
| c | lisinfectant below which there is no know | vn or expected risk to | health. MRDLGs do not | Carbon Dioxide (ppm) | NA NA | 17.5 |
| r | eflect the benefits of the use of disinfect | ants to control microl | oial contaminants. | Conductivity (umhos) | NA NA | 105 |
| | Aillirem per year (mrem/yr) : a measure o | of radiation absorbed | by the body. | Hardness (ppm) | NA NA | 11.3 |

| UNREGULATED CONTAMINANTS | | | | | | | | |
|---------------------------|---------------------|-------------------------|---------------------|-----------------------|---------------------|--|--|--|
| Contaminant | Average Detected | Contaminant | Average Detected | Contaminant | Average Detecter | | | |
| 1,1 – Dichloropropene | ND | Bromobenzene | ND | Isoprpylbenzene | ND | | | |
| 1,1,1,2-Tetrachloroethane | ND | Bromochloromethane | ND | M-Dichlorobenzene | ND | | | |
| 1,1,2,2-Tetrachloroethane | ND | Bromodichloromethane | 6.69 | Methomyl | ND | | | |
| 1,1-Dichloroethane | ND | Bromoform | 0.82 | Metolachlor | ND | | | |
| 1,2,3 - Trichlorobenzene | ND | Bromomethane | ND | Metribuzin | ND | | | |
| 1,2,3 - Trichloropropane | ND | Butachlor | ND | MTBE | ND | | | |
| 1,2,4 - Trimethylbenzene | ND | Carbaryl | ND | N - Butylbenzene | ND | | | |
| 1,3 - Dichloropropane | ND | Chloroethane | ND | Naphthalene | ND | | | |
| 1,3 – Dichloropropene | ND | Chloroform | 17.0 | N-Propylbenzene | ND | | | |
| 1,3,5 - Trimethylbenzene | ND | Chloromethane | ND | O-Chlorotoluene | ND | | | |
| z,z-Dichloropropane | ND | Dibromochloromethane | 0.16 | P-Chlorotoluene | ND | | | |
| 3-Hydroxycarbofuran | ND | Dibromomethane | ND | P-Isopropyltoluene | ND | | | |
| Aldicarb | ND | Dicamba | ND | Propachlor | ND | | | |
| Aldicarb Sulfone | ND | Dichlorodifluoromethane | ND | Sec - Butylbenzene | ND | | | |
| Aldicarb Sulfoxide | ND | Dieldrin | ND | Tert - Butylbenzene | ND | | | |
| Aldrin | ND | Hexachlorobutadiene | ND | Trichlorfluoromethane | ND | | | |

Lead & Copper Monitoring

Central Elmore Water & Sewer Authority completed monitoring requirements for lead and copper in 2022. Thirty-two sites were sampled without exceeding the Action Level Limits for lead or copper. The system will continue to monitor for lead and copper every three years. The next monitoring period for the system will be the period of June - September 2025.

Our monitoring results in 2022 were as follows:

| | | | | EAD & COPPER (TAP V | | | |
|-------------------------------|----------------------|-------------------------------|-------------------------|---------------------------|--------------------------|---|---|
| Contaminant & Unit of MSMT | AL (Action Level) | MCLG (What's the Goal?) | Date Sampled (mo/yr) | 90th Percentile Result | Range Low - High (MD) | No. of Sampling Sites Exceeding the AL | Major Sources |
| Copper (ppm) | 1.3 | 1.3 | June | 0.0732 ppm | 0.0082 - 0.0867 | 0 | Corrosion of household plumbing |
| Lead (ppb) | 15 | 0 | 2022 | 0.44 ppb | ND - 1.4 | 0 | systems; Erosion of natural deposits |

As required by ADEM, we conducted and prepared a Lead Service Line Inventory during 2024. Our findings were:

| | | | ARY |
|------|-------------------|----------|------------------------|
| | DOTAL SERVICE LIN | TES | 15,011 |
| Lead | Galvanized | Non-Lead | Lead Status Unknown |
| 0 | 0 | 15.011 | 0 |

Corrosion of pipes, plumbing fittings and fixtures may cause metals, including lead and copper, to enter drinking water. To assess corrosion of lead and copper, CEW&SA conducts tap sampling for lead and copper at selected sites every three years.

Also, CEW&SA is required to sample for lead in schools and licensed child care facilities as requested by the facility. Please contact your school or child care facility for further information about potential sampling results.

The complete Lead sampling data, Service Line Inventory Report, and any information on replacement plans for Lead, Galvanized, or Unknown service lines are available for review in our office.

| Atuminum (ppm) | 0.05100.2 | ND |
|--------------------------------------|---------------|--------|
| Chloride (ppm) | 250 | 11.6 |
| Color (color units) | 15 | ND |
| Copper (ppm) | 1.0 | 20.1 |
| Corrosivity | Non-corrosive | ND |
| Fluoride (ppm) | 2.0 | ND |
| Foaming agents MBAS (ppm) | 0.5 | 2.4 |
| Iron (ppm) | 0.3 | ND |
| Manganese (ppm) | 0.05 | 7.7 |
| Odor (threshold odor number) | 3 | ND |
| pH (std units) | 6.5 - 8.5 | 12.8 |
| Silver (ppm) | 0.1 | 34 |
| Sulfate (ppm) | 250 | 2.5 |
| Total Dissolved Solids (ppm) | 500 | 155 |
| Zinc (ppm) | 5 | 0.0015 |
| Alkalinity, Total (as CA, Co3) (ppm) | NA NA | 20 |
| Caltium, as Ca (ppm) | NA NA | 2.7 |
| Carbon Dioxide (ppm) | NA NA | 17.5 |
| Conductivity (umhos) | NA NA | 105 |
| Hardness (ppm) | NA NA | 11.3 |
| Magnesium (ppm) | NA NA | 1.12 |
| Nickel (ppm) | NA NA | 0.001 |
| Sodium (ppm) | NA NA | 16.6 |

MCL TT. or MRDL

| PFAS - 2024 | | | |
|---|--------------|--------------------------------------|--------------|
| PFAS Contaminants (ppb) | Max Detected | PFAS Contaminants (ppb) | Max Detected |
| nCI-PF3OUdS (n-chloroeicosafluoro-3oxaundecane-1-sulfonic acid) | ND | Perfluorononanoic acid - PFNA | 0.000001080 |
| 9CI-PF3ONS (9-chlorohexadecafluoro-30xanone-1-sulfonic acid) | ND | Perfluorooctanesulfonic acid - PFOS | 0.000002260 |
| ADONA (4,8-dioxa-3H-perfluorononanoic acid) | ND | Perfluorooctanoic acid - PFOA | 0.000002680 |
| HFPO-DA (Hexafluoropropylene oxide dimer acid) | ND | Perfluorodecanoic acid - PFDA | ND |
| NEtFOSAA (N-ethyl perfluorooctanesul fonamidoacetic acid) | ND | Perfluorododecanoic acid - PFDoA | 0.000000659 |
| NMeFOSAA (N-methyl perfluorooctanesulfonamidoacetic acid) | 0.000000276 | Perfluorohexanoic acid - PFHxA | 0.000000960 |
| Perfluorobutanesulfonic acid - PFBS | 0.000001260 | Perfluorotetradecanoic acid - PFTeDA | 0.000001280 |
| Perfluoroheptanoic acid - PFHpA | 0.000000749 | Perfluorotridecanoic acid - PFTrDA | 0.000000921 |
| Perfluorohexanesulfonic acid - PFHxS | 0.000000598 | Perfluoroundecanoic acid - PFUnA | 0.000000366 |

er- and polyfluoroalkyl substances (PFAS) are a group of man-made chemicals that have properties useful in the manufacture of nonstick pokware, stain-resistant carpet and textiles, firefighting foams, food wrappers, and many more industrial and consumer applications. These chemicals, which have been produced in the United States since the early 1940s, are very persistent in the environment.